Pape	er Code				
Num	lber: 8483		015 (A)	Roll No	
CHEMISTRY PAPE GROUP-I		— INTERMEDIA PER-II (NEW SC	TE PART-II (12 th HEME) <u>JECTIVE</u>	LASS) TIME ALLOWED: 20 Minutes MAXIMUM MARKS: 17	
Note: You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that circle in front of that question number. Use marker or pen to fill the circles. Cutting or filling two or more circles will result in zero mark in that question. Attempt as many questions as given in objective type question paper and leave others blank. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER. Q.No.1					
(1)		he manufacture of synth			
	(A) Formic acid	(B) Oxalic acid	(C) Carbonic acid	(D) Acetic acid	
(2)		made from acrylonitrile			
	(A) PVC	(B) Rayon fibre	(C) Acrylic fibre	(D) Polyester fibre	
(3)	Phosphorus helps	-			
	(A) Root	(B) Leaf	(C) Stem	(D) Seed	
(4)	Ecosystem is a sm				
	(A) Lithosphere	(B) Hydrosphere	(C) Atmosphere	(D) Biosphere	
(5)	Mark the correct statement:- (A) Na^+ is smaller than Na atom				
	(B) Na^+ is larger	than Na atom (C) $C\ell^{2}$	is smaller than $C\ell$ a	tom (D) $C\ell^-$ and $C\ell$ are equal in size	
(6)	The oxides of Bery	yllium are:-			
	(A) Acidic	(B) Basic	(C) Amphoteric	(D) None of these	
(7)		ns an ion with charge +3			
	(A) Beryllium	(B) Aluminium	(C) Carbon	(D) Silicon	
(8)	Oxidation of NO				
	(A) $N_2 O$	(B) $N_2 O_3$	(C) $N_2 O_4$	(D) $N_2 O_5$	
(9)) Hydrogen bonding is the strongest between the molecules of:-				
	(A) HF	(B) <i>HCℓ</i>	(C) HBr	(D) <i>H I</i>	
(10)	0) is a typical Transition metal.				
	(A) <i>Sc</i>	(B) <i>Y</i>	(C) <i>Ra</i>	(D) <i>Co</i>	
(11)) The linear shape is associated with set of hybrid orbitals.				
	(A) <i>sp</i>	(B) sp^2	(C) sp^3	(D) dsp^2	
(12)	Formula of Chloro	oform is:-			
	(A) $CH_3C\ell$	(B) $CC\ell_4$	(C) $CH_2C\ell_2$	(D) $CHC\ell_3$	
(13)	Benzene cannot u	Benzene cannot undergo:-			
	(A) Substitution re	eactions (B) Addition r	eactions (C) Oxidation	n reactions (D) Elimination reactions	
(14)	(14) For mechanisms, the first step involved is the same.				
	(A) $E1$ and $E2$	(B) E2 and $S_N 2$	(C) $S_N 1$ and $S_N 2$	(D) E1 and $S_N 1$	
(15)	compound is called a universal solvent.				
		(B) <i>CH</i> ₃ <i>OH</i>		(D) $CH_3 - O - CH_3$	
(16)	_	tion is not given by:-	~ 2 5		
(10)	(A) Formaldehyde		(C) Benzaldehyde	(D) Trimethyl acetaldehyde	
 (17) reacts with both of Aldehydes and Ketones. 					
(A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's rea				reagent (D) Benedict's reagent	

23(NEW SCHEME)(Obj)(**PP**)-2015(A)-9000 (MULTAN)