	MISTRY PA UP-II	PER-II (NEW SC	CHEME) <u>JECTIVE</u>	TIME ALLOWED: 20 Minutes MAXIMUM MARKS: 17
				B, C and D. The choice which you
think i Cuttin as give BUBB Q.No.1	is correct, fill that ig or filling two or en in objective typ LES are not filled I	t circle in front of that or more circles will resul be question paper and lo d. Do not solve questio	question number. Us t in zero mark in tha eave others blank. N	e marker or pen to fill the circles. t question. Attempt as many questions o credit will be awarded in case
(1)	Acetic acid is man	-	(0) 0	
(2)	(A) Distillation	(B) Fermentation	. ,	(D) Esterification
(2)		een Fat and NaOH is ca		
(2)	(A) Esterification	, , ,	(C) Fermentation	(D) Saponification
(3)		ents are needed for the ho		
	(A) N, S, P	(B) N, Ca, P	(C) N, P, K	(D) N, K, C
(4)	The pH range of			
		(B) $6.5 - 6$		
(5)		the size of atoms, which		
	(A) $Mg > Sr$	(B) $Ba > Mg$	(C) $Lu > Ce$	(D) $C\ell > I$
(6)	does not belong to Alkaline-earth metals.			
	(A) Be	(B) <i>Ra</i>	(C) Ba	(D) Rn
(7)	metal is used in the thermite process because of its reactivity.			
	(A) Iron	(B) Copper	(C) Aluminum	(D) Zinc
(8)	The brown gas formed, when metal reduces HNO ₃ , is:-			
	(A) N_2O_5	(B) N_2O_3	(C) <i>NO</i> ₂	(D) <i>NO</i>
(9)	is used as a cooling medium for nuclear reactors.			
	(A) Ne	(B) <i>He</i>	(C) Ar	(D) <i>Kr</i>
(10)	Coordination number of Pt in $\left[PtC\ell(NO_2)(NH_3)_4\right]$ is:-			
	(A) 2	(B) 4	(C) 1	(D) 6
(11)		ol, the tertiary carbon is b	,	、
()	(A) Two H – atoms (B) Three H – atoms (C) One H – atom (D) No H – atom			
(12)	gas is used for artificially ripening of fruits.			
	(A) Methane	(B) Ethane	(C) Ethyne	(D) Propyne
(13)	The electrophile in aromatic sulphonation is:-			
	(A) H_2SO_4	(B) HSO_4^{-1}	(C) SO_3	(D) SO_3^{+1}
(14)	Elimination bimo	lecular reactions involve	· :-	
	(A) 1 st order Kinetics (B) 2 nd order Kinetics (C) 3 rd order Kinetics (D) Zero order Kinetics			
(15)		wis concept ethers behav		
	(A) Acid	(B) Base	(C) Electrophile	(D) Acid as well as a base
(4.6)	The Carbon atom	of a carbonyl group is _	hybridized.	
(16)				- 2
(16)	(A) <i>sp</i>	(B) sp^2	(C) sp^3	(D) dsp^2
(16) (17)	(A) sp will react	(B) sp^2 with both Aldelyde and I		(D) dsp^2